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This text presents a unified and up-to-date discussion of the role of atomic and molecular orbitals in chemistry, from the quantum mechanical foundations to the recent developments and applications. The discussion is mainly qualitative, largely based on symmetry arguments. It is felt that a sound mastering of the concepts and qualitative interpretations is needed, especially when students are becoming more and more familiar with numerical calculations based on atomic and molecular orbitals. The text is mathematically less demanding than most traditional quantum chemistry books but still retains clarity and rigour. The physical insight is maximized and abundant illustrations are used. The relationships between the more formal quantum mechanical formalisms and the traditional chemical descriptions of chemical bonding are critically established. This book is of primary interest to undergraduate chemistry students and others taking courses of which chemistry is a significant part. These Collections Of The Official Past Papers Of The Gce O Level Examinations From The University Of Cambridge International Examinations Has Been Developed For Students Of Gce O Level. These Books Will Act As Tools For Preparation And Revision For Students. These Books Have An Edited Answer Guide For Each Paper Based On The Marks Scheme Written By Cie Principal Chemistry for the IB Diploma, Second edition, covers in full the requirements of the IB syllabus for Chemistry for first examination in 2016. Fully updated and matched to the Cambridge syllabus, this stretching Student Book is trusted by teachers around the world to support advanced understanding and achievement at IGCSE. The popular, stretching approach will help students to reach their full potential. Written by experienced authors, this updated edition is full of engaging content with up-to-date examples to cover all aspects of the Cambridge syllabus. The step-by-step approach will lead students through the course in a logical learning order building knowledge and practical skills with regular questions and practical activities. Extension material will stretch the highest ability students and prepare them to take the next step in their learning. Practice exam questions will consolidate student understanding and prepare them for exam success. You will also receive free access to extra support online, including practice exam questions, revision checklists and advice on how to prepare for an examination. Endorsed by WJEC, and written by a team of experienced senior examiners this is the only study and revision guide that precisely matches the WJEC AS Chemistry course. It contains essential course notes, revision advice, and examiner tips on how to boost grades as well as a Q&A section with model student answers, examiner commentaries and marks. Fully revised and updated content matching the Cambridge International AS & A Level Chemistry syllabus (9701). Endorsed by Cambridge International Examinations, the Second edition of the AS/A Level Chemistry Coursebook comprehensively covers all the knowledge and skills students need for AS/A Level Chemistry 9701 (first examination 2016). Written by renowned experts in Chemistry, the text is written in an accessible style with international learners in mind. The Coursebook is easy to navigate with colour-coded sections to differentiate between AS and A Level content. Self-assessment questions allow learners to track their progression and exam-style questions help learners to prepare thoroughly for their examinations. Contemporary contexts and applications are discussed throughout enhancing the relevance and interest for learners. ISBN: 9781741252996 AUTHOR: Jim Stamell RRP: \$39.95 PAGES: 428 pp. SPECIFICATION: Softcover, perfect bound, 280 mm x 210 mm STATUS: New edition PUBLICATION DATE: April 2008 The EXCEL HSC Chemistry guide is directly linked to the syllabus with every single dot point of the HSC Chemistry syllabus appearing in the margin of the book. You can write in the guide, so your study is focused and your notes are structured. This guide comes in a brand new format that makes even better use of your study time! up-to-date coverage of the core topics plus 3 Option topics: Industrial Chemistry, Shipwrecks, Corrosion and Conservation and Forensic Chemistry. this guide is organised just like the HSC syllabus, so the students learn to section (the theoretical part) is under routine headings and the students section (the practical part) is under headings like First-hand/Second-hand and Investigations and Problem Solving - %this way you will be able to see at a glance what the theoretical and practical work is! all main headings in each chapter (1. 1, 2. 1, etc.) are directly from the syllabus, word for word %this way you can easily match the Excel guide to the syllabus! an alphabetical list of all the key definitions and concepts you should know from each chapter %an efficient way of learning all the definitions in one go! chapter syllabus checklist with every single dot point listed in checklist form for each chapter %a fantastic way of testing that you know all the work ! hundreds of key concept questions with answers %questions that test your recall of knowledge in each chapter. HSC-type questions for every section in each chapter with clock icons to tell you how much time you will have to answer the questions in the HSC %this way you can test yourself on HSC-type questions under HSC-type time pressure! an examiner maximiser feature, ticks to show the mark distribution and answers to all HSC-type questions - %all you need to answer HSC-type questions! two sample HSC papers with an examiner maximiser feature plus answers %not one but two up-to-date sample papers ! the Excel syllabus summary notes: a detachable section at the end of the guide, where every single dot point of each chapter is summarised for you% - a comprehensive and compact summary of the whole course in 32 pages! Alchemy can't be science--common sense tells us as much. But perhaps common sense is not the best measure of what science is, or was. In this book, Bruce Moran looks past contemporary assumptions and prejudices to determine what alchemists were actually doing in the context of early modern science. Examining the ways alchemy and chemistry were studied and practiced between 1400 and 1700, he shows how these approaches influenced their respective practitioners' ideas about nature and shaped their inquiries into the workings of the natural world. His work sets up a dialogue between what historians have usually presented as separate spheres; here we see how alchemists and early chemists exchanged ideas and methods and in fact shared a territory between their two disciplines. Distilling Knowledge suggests that scientific revolution may wear a different appearance in different cultural contexts. The metaphor of the Scientific Revolution, Moran argues, can be expanded to make sense of alchemy and other so-called pseudo-sciences--by including a new framework in which "process can count as an object, in which making leads to learning, and in which the messiness of conflict leads to discernment." Seen on its own terms, alchemy can stand within the bounds of demonstrative science. Exam Board: SQA Level: Higher Subject: Chemistry First Teaching: 2018, First Exam: 2019 Two books in one! Combining a revision guide, a full set of practice test papers, this fantastic resource is all you need to revise for the new 2019 exam. The revision guide: - covers all of the topics in the new CfE Higher Chemistry curriculum, broken down into manageable chunks for easy revision- clearly explains key concepts, research evidence and real-life applications- contains Quick Tests to let students check their knowledge and understanding as they go along The practice test papers: - are in the format and the style of the SQA exam, giving students an opportunity to practice taking the new 2019 Higher Chemistry exam Marking instructions and sample answers are provided online, so students can check their progress. This edition of our successful series to support the Cambridge IGCSE

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Because our own historical moment continues to be indebted to romanticism, such a shift in understanding prompts a rethinking in our ideas of the interrelation of literature, philosophy, and science."--Jacket. Where Stonehenge's giant bluestones come from? Was the fall of the Roman Empire hastened by lead poisoning? How did amber get from the Baltic to Belize? In exploring these and other historical enigmas, Joseph Lambert expertly details the rich insights into ancient life that chemistry alone can provide. Using cutting-edge scientific methods such as radiocarbon dating, DNA analysis, and elemental fingerprinting, acclaimed chemist Joseph Lambert expertly details the rich insights into ancient life that chemistry alone can provide. He shows, for example, how investigators today can determine the diet of prehistoric Europeans, the geographical origin of the marble in a Greek statue, or the reason why the Liberty Bell cracked. He uses nuclear magnetic resonance spectroscopy to reconstruct ancient trade routes, and X-ray diffraction, among other methods, to compare the color palettes of the Mesopotamians and Egyptians (the latter were apparently much more flamboyant). He explains how chemical analysis of DNA can be used to sort out human lineages and migratory patterns—demographic trends that affected, in turn, everything from language to the spread of disease. Chemistry takes center stage in this fascinating book, proving that it is not just an analyst of culture, it stands as one of its primary creators. Lambert offers us a unique glimpse into a form of technical progress hitherto unappreciated: the ever-increasing ingenuity of the Human race, as seen through the prism of its evolving chemical sophistication. We discover how primitive chemistry was initially used by ancient people as a tool to improve their daily lives, a feat that was achieved by reworking molecules of clay into pottery and minerals into metal alloys, and by turning grains into beer and pitch into sealants. By documenting the way ancient people manipulated their environment chemically, Lambert further refines the distinguishing feature of our species. Early humans were more than tool-makers. They were molecular transformers. The scientific and economic importance of the high-temperature reactions of hydrocarbons in both the presence and absence of oxygen cannot be overemphasized. A vast chemical industry exists based on feedstocks produced by the controlled pyrolysis of hydrocarbons, while uncontrolled combustion in air is still among the most important sources of heat and mechanical energy. The detonation and explosion of hydrocarbon-oxidant mixtures can however, be a highly dangerous phenomenon which destroys lives and equipment. In order that control can be exerted over combustion processes, a complete description of hydrocarbon oxidation and pyrolysis is required. A major contribution to this is an understanding of the unstable intermediates involved and their reactions. The aim of this book is to review our knowledge of the chemistry of hydrocarbon combustion and to consider the data which are available for relevant reactions. Chapter 1 describes early studies in which the apparent complexity of the chemistry was established and the type of information required for a better understanding was defined. Experimental studies of the overall process which were carried out with the aim of establishing the sequence of stable chemical intermediates and some of the unstable species are described in Chapter 2. The limited nature of the information thus obtained showed that independent studies of individual reactions involving the unstable species were required. In Chapter 3

investigations specifically aimed at the determination of the kinetics of elementary reactions are discussed. This Practice Book supports the existing and bestselling edition of IGCSE Chemistry Student's Book. - The perfect resource to use throughout the course to ensure you learn the topics and practise the content of the Cambridge IGCSE syllabus. - Contains a wealth of levelled questions, including Stretch and Challenge for higher ability students. - Plenty of exam-style questions and actual exam questions from past Cambridge exam papers for exam success. Exam Board: WJEC Level: GCSE Subject: Chemistry First Teaching: September 2016 First Exam: June 2018 Welsh edition. 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The new technology program reinforces the approach of the text and provides a complete solution for teaching and learning. The Eighth Edition retains the hallmark pedagogical features of the text and builds upon the conceptual focus and art program. Students also benefit from online homework in the technology program, which features an extensive database of questions drawn from the text. Endorsed by Cambridge International Examinations Covers the entire syllabus for Cambridge International Examinations' International AS and A Level Chemistry (9701). It is divided into separate sections for AS and A Level making it ideal for students studying both the AS and the A Level and also those taking the AS examinations at the end of their first year. - Explains difficult concepts using language that is appropriate for students around the world - Provides practice throughout the course with carefully selected past paper questions at the end of each chapter - Lets students work through problems at their own pace, with a free Revision and practice CD which includes interactive tests, answers to selected questions, additional activities, and a glossary - Answers to the questions in the Student's Book can be found on the Teacher's CD In this history of materials, the authors link chemical science with chemical technology, challenging our current understandings of objects in the history of science and the distinction between scientific and technological objects. They further show that chemists' experimental production and understanding of materials changed over time, first in the decades around 1700 and then around 1830, when mundane materials became clearly distinguished from true chemical substances. The many-faceted efforts to understand the structure and interactions of atoms over the past hundred years have contributed decisively and dramatically to the explosive development of physics. There is hardly a branch of modern physical science that does not in some seminal way rely on the fundamental principles and mathematical and experimental insights that derive from these studies. In particular, the drive to understand the singular features of the hydrogen atom--simultaneously the archetype of all atoms and the least typical atom--spurred many of the twentieth century's advances in physics and chemistry. This book gives an in-depth account of the author's own penetrating experimental and theoretical investigations of the hydrogen atom, while simultaneously providing broad lessons in the application of quantum mechanics to atomic structure and interactions. A pioneer in the combined use of atomic accelerators and radiofrequency spectroscopy for probing the internal structure of the hydrogen atom, Mark Silverman examines the general principles behind this far-reaching experimental approach. Fast-moving protons are directed into gas or foil targets from which they capture electrons to become hydrogen atoms moving uniformly at very high speeds. During their rapid passage through the spectroscopy chamber of the atomic accelerator, these atoms reveal by the light they emit fascinating details of their internal configuration and the interactions that created them. Silverman examines the effects of radiofrequency fields on the hydrogen atom clearly and systematically, explaining the details of these interactions at different levels of complexity and refinement, each level illuminating the physical processes involved from different and complementary perspectives. Readers interested in diverse areas of physics and physical chemistry will appreciate both the theoretical and practical implications of Silverman's studies and the personal style with which he relays them. This is a work of not only an outstanding research physicist, but a fine teacher who understands how curiosity underlies all science. 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